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Heat And Its Measurement Answers

Explain the terms heat and temperature. Heat is a form of energy that follows

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from a body of higher temperature to one of lower temperature. Temperature is a degree of hotness or coldness of a body. 2. What is thermometry? The science of temperature and its measurement is called thermometry. 3. Discuss various scales of thermometer.

Heat And Temperature Grade 11

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Physics Question Answer ...

Heat & Its Measurement 1. How many Joules of heat are given off when 5.00 grams of water cool from 75 °C to 25 °C? $Q = m \cdot C_p \cdot \Delta T$ $Q = 5.00 \text{ g} \cdot 4.18 \text{ J/g}\cdot\text{°C} \cdot (75 - 25 \text{ °C}) = 1045 \text{ J}$ 2. If 575 Joules of heat are added to 45.0 grams of water at 25.0 °C, what will the new temperature be? $Q = m \cdot$

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Class 7 Science Chapter 4 MCQ (Multiple Choice Questions ...

Answer: Heat is a form of energy which

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produces a sensation of hotness or coldness. It is measured in Joules (J) or in Calories. Heat energy flows from one body to another due to temperature difference between the two. Heat is an internal energy of the substance due to molecular motion of the substance.

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Heat, as distinct from temperature, can only be measured using a medium, ie a substance of known specific heat. Then by measuring its temperature the

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amount of heat can be calculated. Water is the ...

What type of measurement is used to measure heat? - Answers

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Worksheet 5 Skeletal System,

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Heat is defined as energy in transit from a warmer body to a cooler body. In SI, its unit of measurement is the joule (J).

Measurement of body heat is called what? - Answers

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The answer is (a) is a better conductor of

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heat than the wooden tile. 2. A beggar wrapped himself with a few layers of newspaper on a cold winter night. This helped him to keep himself warm because (a) friction between the layers of newspaper produces heat. (b) air trapped between the layers of newspaper is a bad conductor of heat.

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Question 1. In a mercury thermometer, the level of mercury rises when its bulb comes in contact with a hot object.

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The standard unit of measurement of heat is Joules, while that of temperature is Kelvin, but it can also be measured in Celsius and Fahrenheit. Calorimeter is a device, which is used to measure the

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heat. On the other hand, temperature can be measured by thermometer. Heat is represented by 'Q' whereas 'T' is used to represent temperature.

Difference Between Heat and Temperature (with Comparison ...

7. Specific heat is - the specific temperature at which the substance is in

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solid state. the energy needed to increase the temperature of 1 gram of a substance by 1 degree Celsius. the amount of heat conducted in 1 minute. the heat needed to increase the temperature of 1 gallon of water by 1 degree Fahrenheit.

GK Questions on Heat and

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Temperature

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